

# Semiconductor Process Presentation

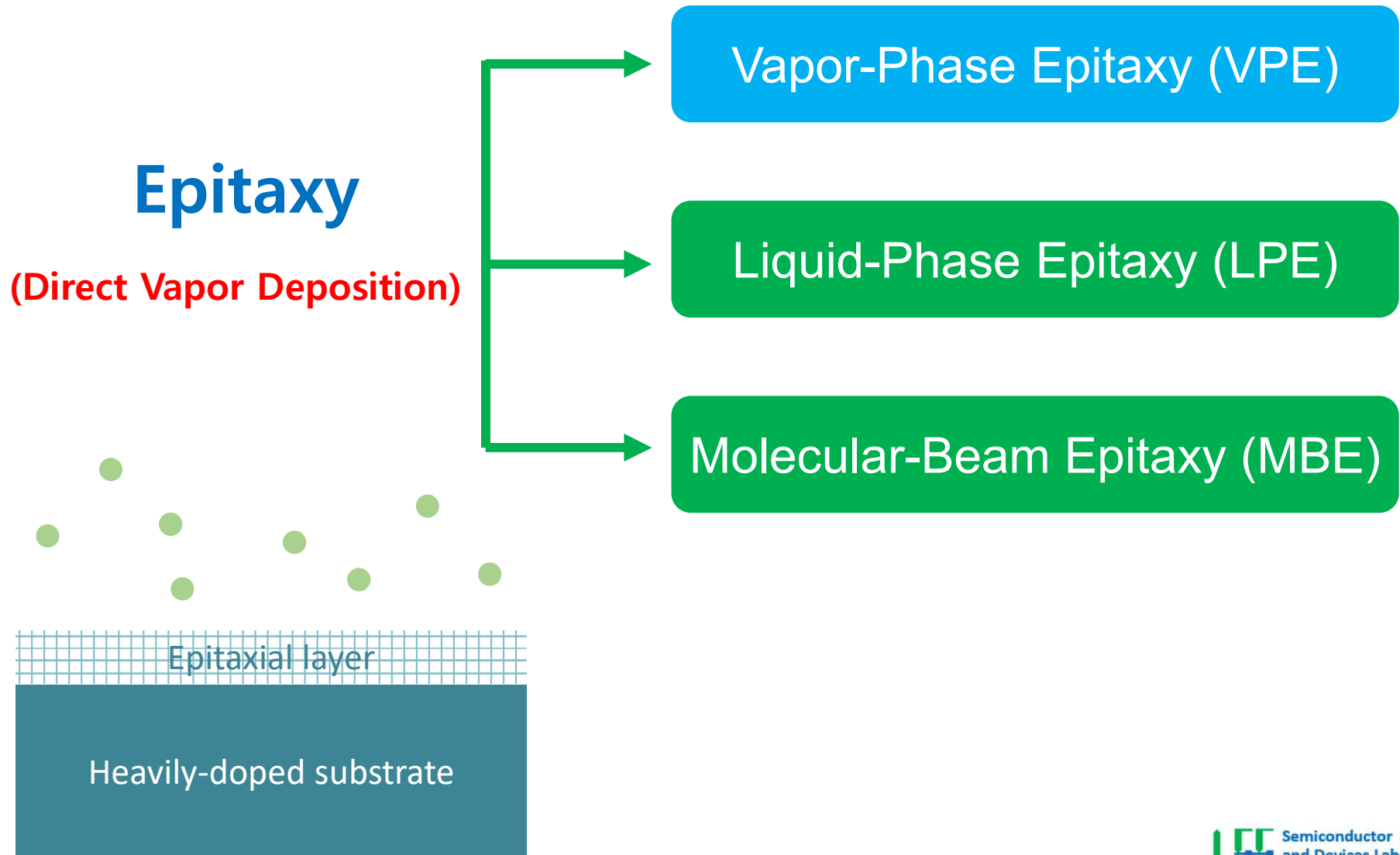
2019.12.05



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# 1. Epitaxy



## 2. Vapor-Phase Epitaxy (VPE)

$$J_S = k_S N_S \quad [\#/m^2 \cdot \text{sec}]$$

$k_S$  is the surface reaction rate constant  $[m/\text{sec}]$

$N_S$  is the surface concentration  $[\#/m^3]$

$$J_g = (\overline{D}_g / \delta) \cdot (N_g - N_S) = h_g \cdot (N_g - N_S) \quad [\#/m^2 \cdot \text{sec}]$$

$\overline{D}_g$  is an effective diffusion constant  $[m^2/\text{sec}]$

$\delta$  is the distance over which the diffusion is taking place

$h_g$  is the vapor phase mass transfer coefficient  $[m/\text{sec}]$

$$N_S = \frac{h_g}{h_g + k_S} \cdot N_g$$

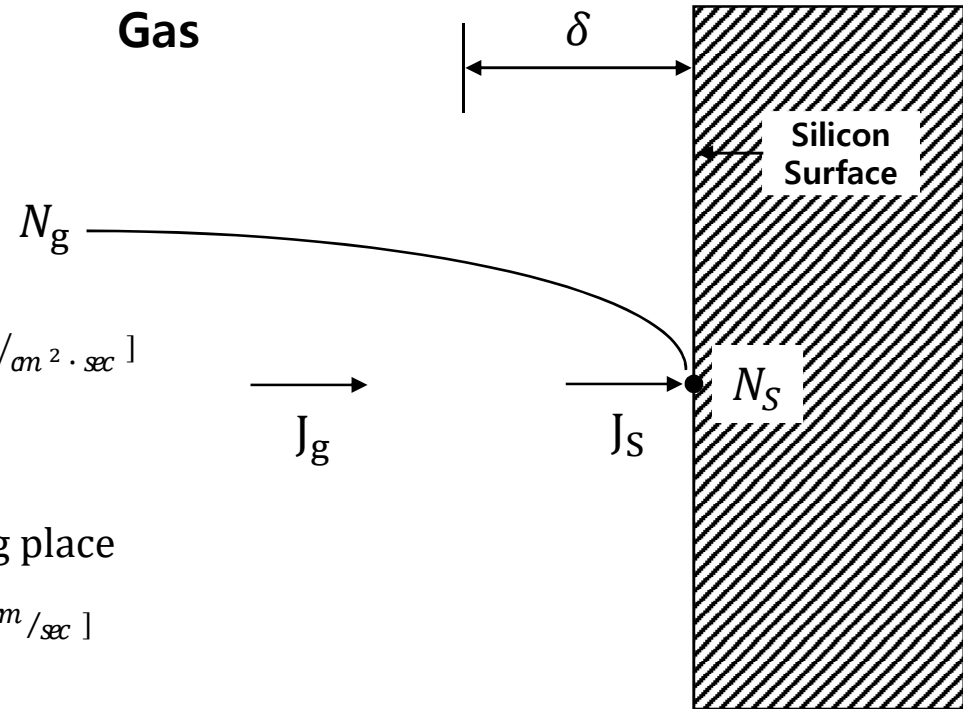
$$J_S = k_S N_S$$

$$v = \frac{J_S}{N} = \frac{k_S h_g}{k_S + h_g} \cdot \frac{N_g}{N}$$

$v$  is growth rate  $[m/\text{sec}]$

$$\left. \begin{array}{l} \\ \end{array} \right\} v = h_g \cdot \frac{N_g}{N} \quad (\text{mass transfer limited, } k_S \gg h_g)$$

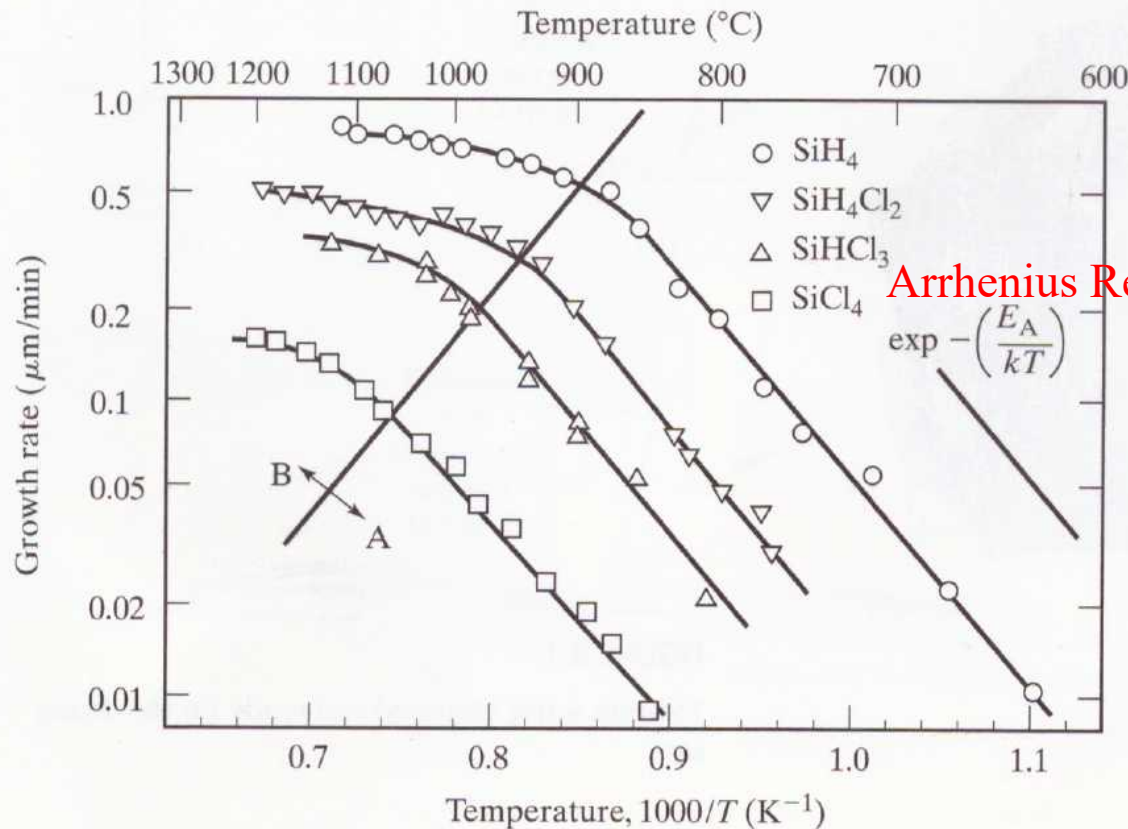
$$\left. \begin{array}{l} \\ \end{array} \right\} v = k_S \cdot \frac{N_g}{N} \quad (\text{surface reaction limited, } h_g \gg k_S)$$



## 2. Vapor-Phase Epitaxy (VPE)

A: Surface-reaction-limited region

B: Mass-Transfer-limited region



Arrhenius Relationship

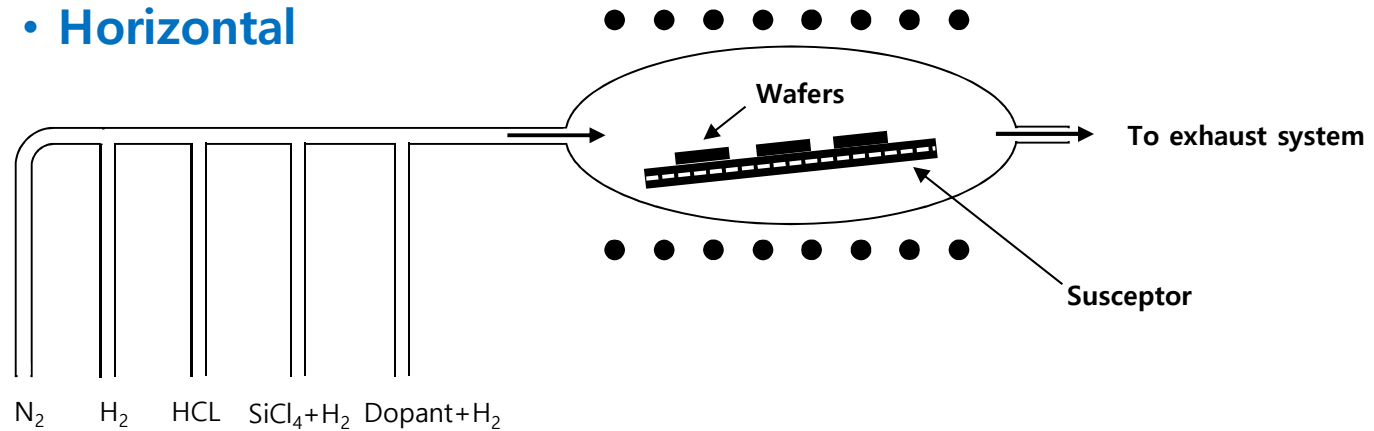
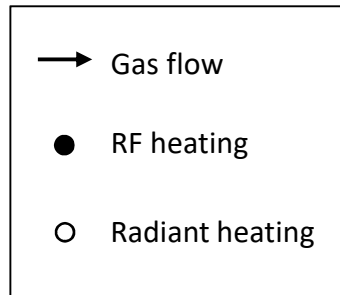
$$\exp\left(-\frac{E_A}{kT}\right)$$

FIGURE 6.10

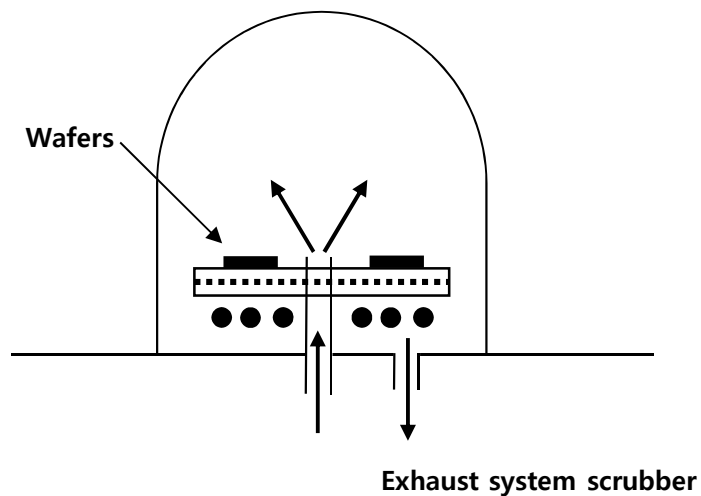
Temperature dependence of the silicon epitaxial growth process for four different sources. The growth rate is surface-reaction-limited in region A and is mass-transfer-limited in region B. Reprinted with permission from Philips Journal of Research from Ref. [3].

# 2. Three Common Types of VPE Reactors

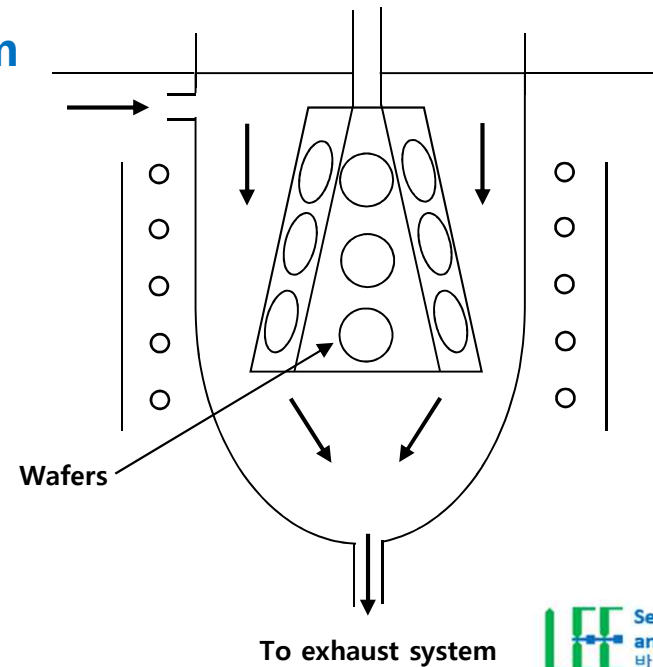
## • Horizontal



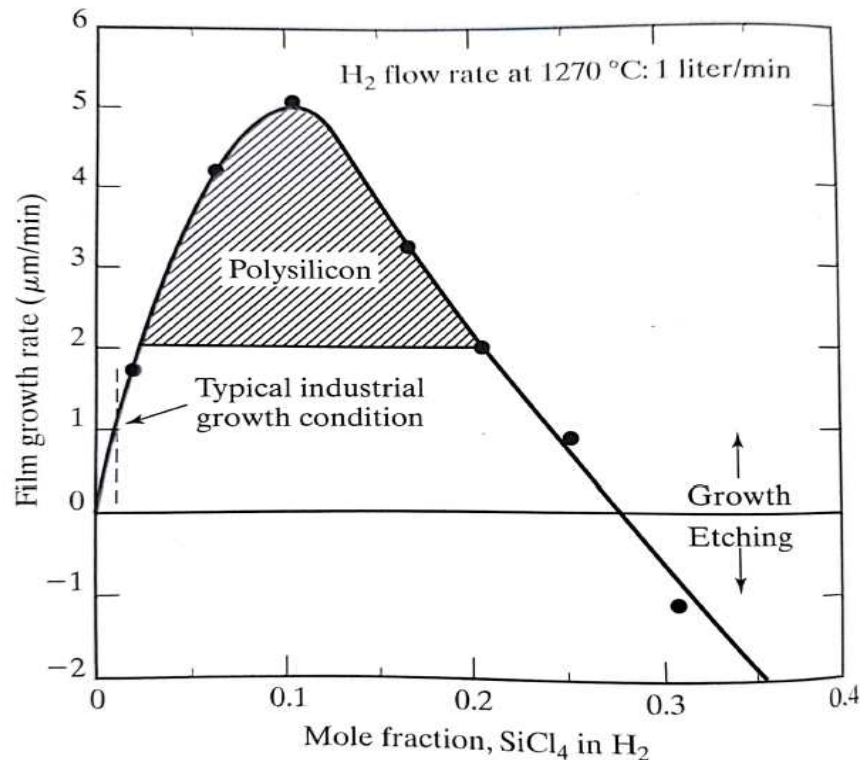
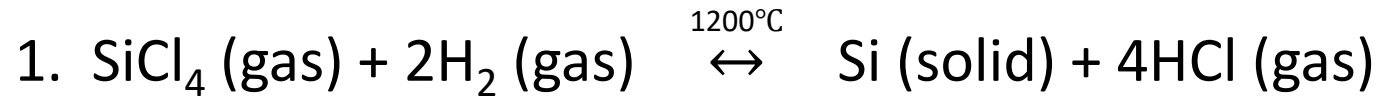
## • Pancake



## • Barrel system



## 2. VPE's Reactions



- The increase of Mole Fraction  
→ Etching
- The increase of Growth Rate  
→ Polysilicon

# 3. Doping of Epitaxial Layers

● :  $\text{SiCl}_4$  &  $\text{H}_2$  /  $\text{SiH}_4$  Enhancing the growth rate

● :  $\text{B}_2\text{H}_6$  (Diborane)

    /  $\text{AsH}_3$  (Arsine)

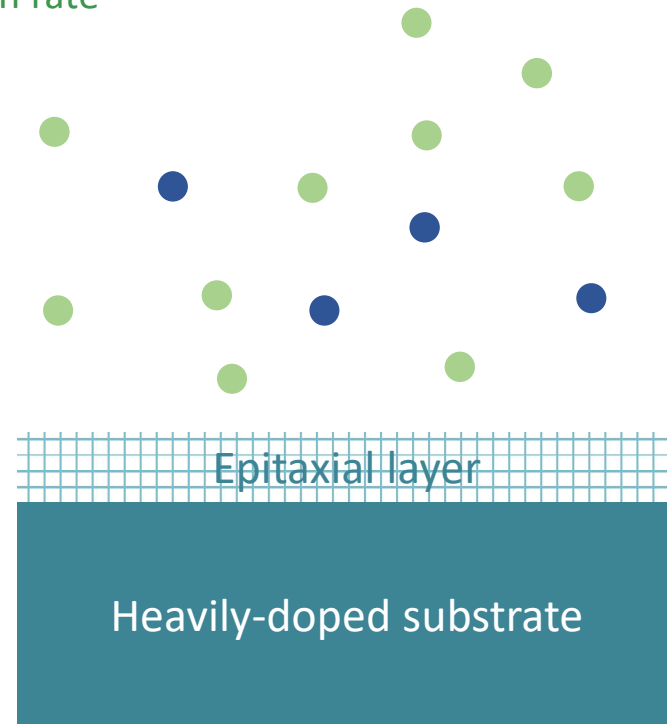
    /  $\text{PH}_3$  (Phosphine) / etc.

→ Gas source of impurities

Slowing down the growth rate

Doping control

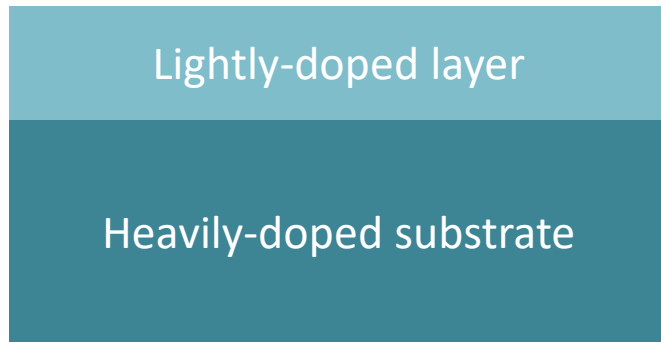
→ by varying the partial pressure of the dopant species



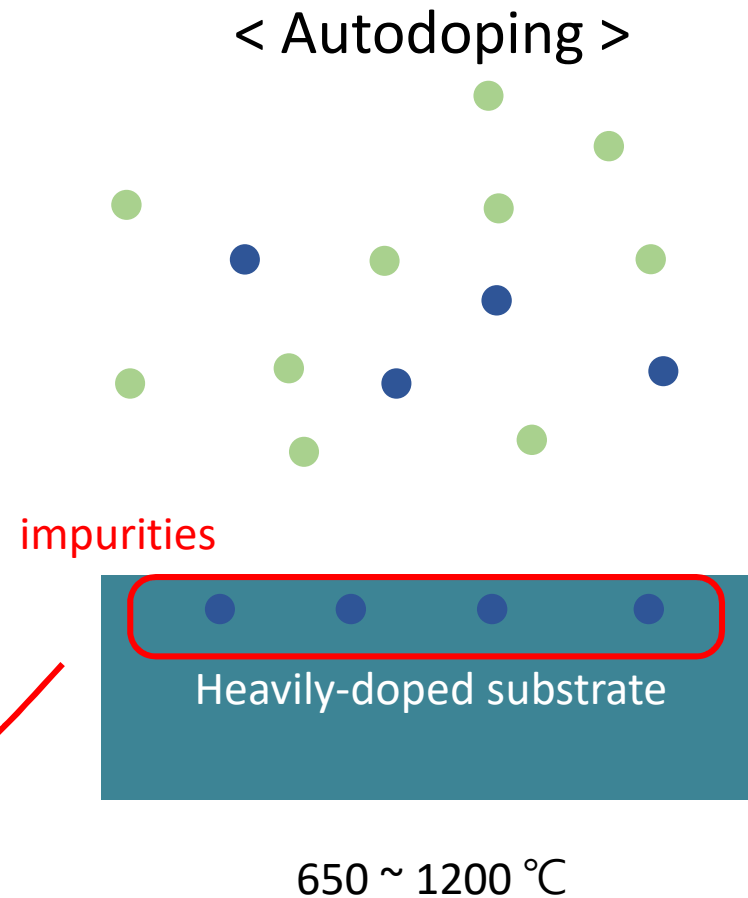
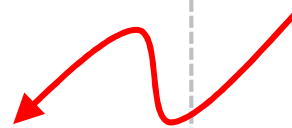
650 ~ 1200 °C



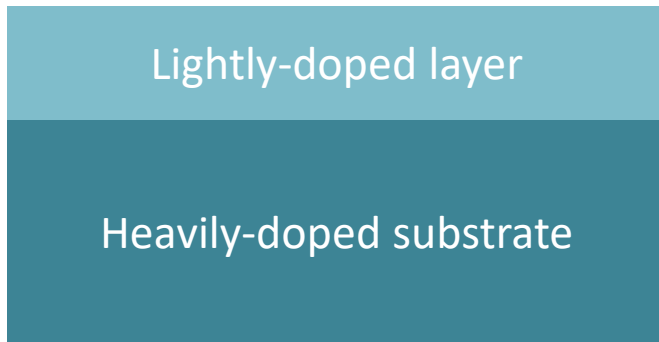
# 3. Doping of Epitaxial Layers



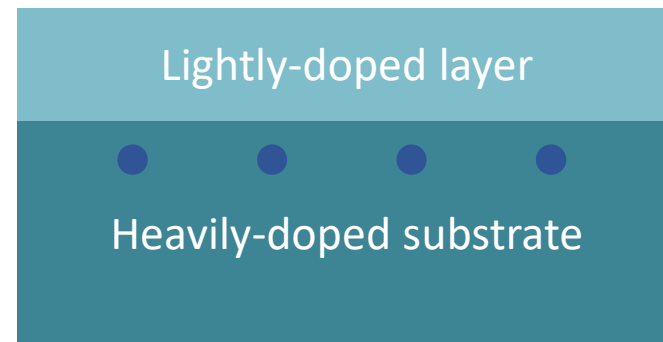
- Evaporate from the wafer
- Liberated by **Cl** etching



# 3. Doping of Epitaxial Layers



< Out-diffusion >

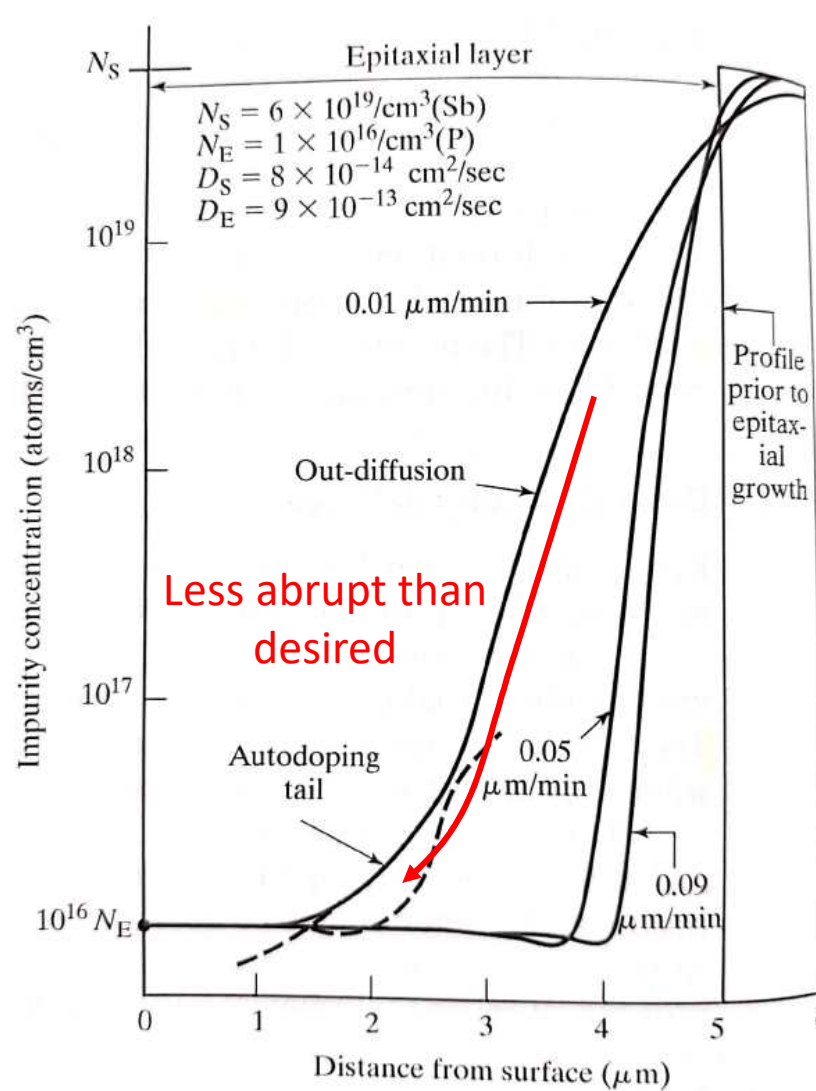


650 ~ 1200 °C

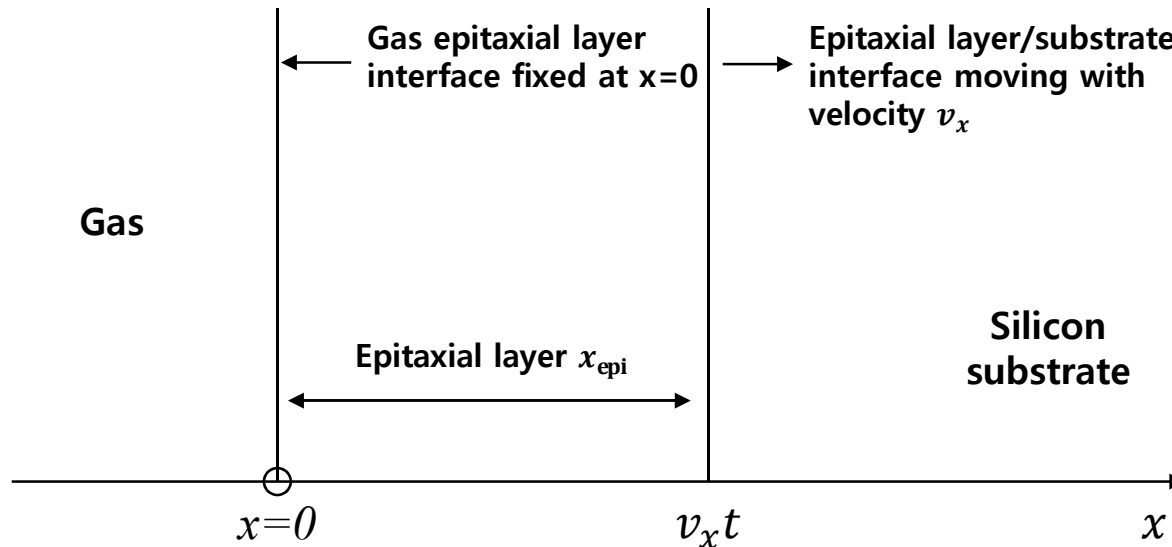
→ A common problem with buried layer

# 3. Doping of Epitaxial Layers

at 1150°C



# 4. Out-Diffusion's Problem



- $D \frac{\partial^2 N}{\partial x^2} = \frac{\partial N}{\partial t} + v_x \frac{\partial N}{\partial x}$

The growth of an undoped epi-layer on a uniformly doped substrate

- $N_1(x, t) = \frac{N_S}{2} \left[ 1 + \operatorname{erf} \frac{x-x_{\text{epi}}}{2\sqrt{D_S t}} \right]$  (B.C.:  $N(x, 0) = N_S = N(\infty, t)$  &  $J_x = (h + v_x)N(0, t)$ )

- $N_2(x, t) = \frac{N_S}{2} \left[ \operatorname{erfc} \frac{x-x_{\text{epi}}}{2\sqrt{D_E t}} + \exp \frac{v_x x}{D_E} \operatorname{erfc} \frac{x+x_{\text{epi}}}{2\sqrt{D_E t}} \right]$  (B.C.:  $N(0, t) = N_E$  &  $N(\infty, t) = 0 = N(x, 0)$ )

- $N(x, t) = N_1(x, t) + N_2(x, t)$

The growth of a doped epi-layer on an undoped substrate

# 4. An Additional Problem : Pattern Shift

